

INTERACTION OF IMMOBILIZED HUMAN PLASMINOGEN AND PLASMIN
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The complex of immobilized human plasminogen (iHPlg) and streptokinase (SK) has a lower catalytic activity in hydrolysis of plasmin-specific substrate S-2251 with a k_{cat}/K_m of $0.005 \mu\text{M}^{-1}\cdot\text{s}^{-1}$, compared to $0.031 \mu\text{M}^{-1}\cdot\text{s}^{-1}$ of immobilized human plasmin (iHPlm) and $0.078 \mu\text{M}^{-1}\cdot\text{s}^{-1}$ of HPlm. The SK in the complex could be removed by acid buffer solution, and the iHPlg will remain catalytically active. iHPlm-SK complex could activate bovine plasminogen (BPlg), whereas iHPlg-SK complex could not. iHPlg could be activated by interaction with a combination of SK and HPlg or urokinase. The activated iHPlg-SK complex had the ability to activate BPlg as iHPlm-SK. The reasonable explanation is that iHPlg was converted to iHPlm in reaction with the combination of HPlg and SK. However, iHPlg was converted to a virgin enzyme in reaction with SK alone and could not activate BPlg. A new modified mechanism for the interaction of HPlg or HPlm with SK was proposed. © 1993 Academic Press, Inc.

Activation of mammalian plasminogen (Plg) to plasmin (Plm), which is the major fibrinolytic enzyme, occurs as a result of cleavage of a peptide bond at Arg⁵⁶⁰-Val⁵⁶¹ in the zymogen molecule (1-3). Several Plg activators, such as streptokinase (SK), urokinase and tissue-type plasminogen activator can catalyze the reaction and are clinically used as thrombolytic agents (4). The bacterial Plg activator SK is unique in that it is a proactivator with no proteolytic activity (5, 6). The true activator is formed as it interacts with some species-specific Plg or Plm (7-11). The interaction between protease active site domain of Plg or Plm and SK could be the main cause for the complex formation (12-16). Previous studies have demonstrated that human plasminogen (HPlg) or human plasmin (HPlm) and SK equimolar complex can activate Plg of other species, such as bovine plasminogen (BPlg), which cannot be activated by SK alone (12, 17-19). The HPlg-SK complex was suggested to be an efficient Plg activator (8, 20). Binding

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Abbreviations used: Plg, plasminogen; Plm, plasmin; HPlg and BPlg, human and bovine plasminogens, respectively; HPlm and BPlm, human and bovine plasmins, respectively; iHPlg and iHPlm, immobilized human plasminogen and plasmin, respectively; SK, streptokinase, and vHPlg, virgin enzyme of human plasminogen.

of SK to HPIg was found to induce the formation of a functional activator active site in the HPIg moiety (7, 8, 19). As a result, a virgin enzyme (vHPIg), a catalytically active form of HPIg with intact Arg⁵⁶⁰-Val⁵⁶¹ peptide bond, was detected as SK was separated from the HPIg-SK complex by acid solution (21).

It has been suggested that HPIg-SK can be converted to HPI_m-SK intramolecularly, although no direct evidence has been provided (21-24). In this study, iHPIg and iHPI_m were used to reinvestigate the mechanism of interaction between SK and PIg or Pl_m. The iHPIg or iHPI_m would limit the interaction between molecules of PIg or Pl_m. The linkage with the solid support would prevent the reactions that require the intermediates containing two HPIg or HPI_m, such as HPIg-SK-HPIg or HPI_m-SK-HPI_m. However, the formation of these intermediates becomes possible if HPIg or HPI_m is added to the solution. Therefore, iHPIg was applied to study the conversion of the HPIg molecule in the HPIg-SK complex to a virgin enzyme or to an HPI_m. The amidolytic activity and the ability to activate BPIg of these complexes were also studied.

MATERIALS AND METHODS

Proteins and enzymes: HPIg was prepared from pooled human plasma by a modification of the Deutsch and Mertz method (25). Forms 1 and 2 of native human Glu-PIg were separated by chromatography on Lys-Sepharose column (26). Form 2 of PIg was used throughout the experiment. HPI_m was prepared by activating HPIg with Sepharose-bound urokinase as previously described (15). BPIg was purified from fresh citrated bovine plasma by the same affinity chromatography method. SK (750,000 IU/mg, from Behringwerke AG, Marburg, W. Germany) was further purified by passing it through a blue-Sepharose CL 6B column (0.9 x 40 cm) to remove serum albumin (17).

Preparation of protein substituted Sepharose: HPIg or HPI_m (6 mg) was coupled to CNBr-activated Sepharose 4B (1 g, from Pharmacia LKB Biotechnology Inc.) in a coupling buffer containing 0.1 M NaHCO₃, 0.5 M NaCl (pH 8.3) at 4°C overnight. The gel was incubated and washed with 1 M ethanolamine in 0.1 M NaHCO₃, 0.5 M NaCl (pH 8.0) to eliminate the residual active groups. The gel was then washed with coupling buffer (pH 8.3) and 0.1 M acetate buffer, 0.5 M NaCl (pH 4.0) alternately and was preserved in activating buffer pH 6.5, containing 0.05 M phosphate, 0.1 M NaCl, 0.02 M lysine, and 25% glycerol.

Protein and enzyme concentration: Protein concentrations were determined spectrophotometrically using the following $\epsilon^{1\%}_{1\text{cm}}$, 280 nm values and molecular weights, respectively: HPIg and BPIg, 17.0 and 84,000 (2, 27, 28); HPI_m, 17.0 and 76,500 (28); SK, 9.5 and 45,000 (29). Active site concentration of HPI_m was determined by the *p*-nitrophenyl-*p*'-guanidinobenzoate burst titration (15, 30). The active concentration of iHPI_m was determined by measuring its amidolytic activity according to a standard HPI_m calibration curve. The amount of protein on the immobilized gel was determined by amino acid analysis. iHPIg or iHPI_m (0.1 g) was washed extensively with 30 ml of distilled water, vacuum-dried, and hydrolyzed in 6 N HCl (Sequanal-grade, Pierce Chemical Co.) at 110°C for 24 h. Amino acid analysis was performed on a Beckman Model 6300 amino acid analyzer.

Amidase parameter measurement: The enzymatic activities of iHPI_m and iHPIg-SK complex were measured with the peptide substrate, NH₂-D-Val-Leu-Lys-*p*-nitroanilide (S-2251), at 37°C and in 0.05 M Tris-HCl, pH 7.4, 0.1 M NaCl. The substrate concentration varied between 0.2 and 4 K_m. At the indicated time, the reaction mixture was rapidly filtered through a plastic syringe with a tight glass-wool plug at the end. The absorbance at 405 nm of the filtrate was measured. The initial rate and substrate concentration data were analyzed on a Lineweaver-Burk plot. The $\epsilon^{1\text{M}}$ at 405 nm employed for *p*-nitroanilide was 9559.

Preparation of iHPIIm-SK complex, iHPIg-SK complex, and virgin enzyme (vHPIg):

IHPIIm (4 μ M) or iHPIg (4 μ M) was interacted with equimolar SK in 0.5 ml activating buffer (pH 6.5) at 25°C for 30 min with continuous stirring. The iHPIg-SK complex was an amidolytically active virgin enzyme. The iHPIg-SK or iHPIIm-SK complex was packed into a small plastic column and washed with 20 ml of activating buffer (pH 6.5) followed by 20 ml of 0.05 M phosphate, 0.1 M NaCl, 0.02 M lysine, 25% glycerol buffer (pH 2.2) at 4°C to remove SK. Protein labeled with 125 I was used to monitor the removal of SK and HPIg added in the solution. The gels were resuspended in 0.5 ml of activating buffer for enzyme activity assay.

125 I-labeling of protein: SK or HPIg (100 μ l) at a concentration of 1.0 mg/ml was incubated with Na 125 I (100 μ Ci) and slightly agitated at room temperature in an eppendorf tube previously coated with 100 μ l Iodo-gen (0.08 mg/ml) (Sigma) for 8 min. After incubation, the reaction mixture was applied on a PD-10 column (Pharmacia, Sweden) and eluted with 0.01 M phosphate buffer (pH 7.2). The labeled protein was collected after discarding the void volume.

Amidolytic activity measurement: To measure the amidolytic activity of the immobilized enzyme, 30 μ l of the gel suspension (containing immobilized enzymes at a final concentration of 0.24 μ M) was added into 0.5 ml of 0.05 M Tris-HCl (pH 7.4) buffer containing 0.5 mM peptide substrate S-2251 and 0.1 M NaCl at 37°C. The rate increment of absorbance at 405 nm of the supernatant was recorded.

BPIg activator activity measurement: BPIg (5 μ M) was incubated with various preparations of iHPIIm-SK or iHPIg-SK in 0.5 ml of activating buffer at 37°C. At the time indicated, 50 μ l aliquot of the reaction mixture was pipetted to a cuvette containing 0.5 mM peptide substrate (S-2251) in 0.05 M Tris-HCl (pH 7.4), 0.1 M NaCl. The initial rate of increment of absorbance at 405 nm was recorded. The reaction without BPIg was used as control.

Activation of iHPIg: IHPIg before and after washing with 20 ml of cold 0.05 M phosphate, 0.1 M NaCl, 0.02 M lysine, 25% glycerol buffer (pH 2.2) was incubated with catalytic amount of urokinase, SK, or equimolar HPIIm-SK in activating buffer (pH 6.5) at 30°C. At the time indicated, 25 μ l of the gel suspension was added to 0.5 ml of 0.05 M Tris-HCl (pH 7.4) containing 0.5 mM S-2251 and 0.1 M NaCl at 37°C and the absorbance at 405 nm of the supernatant was recorded. The initial rate was calculated.

SDS-polyacrylamide gel electrophoresis: The basic techniques of gel electrophoresis were done as described by Weber and Osborn (31).

Reagents: S-2251, urokinase, and *p*-nitrophenyl-*p'*-guanidinobenzoate were obtained from Sigma. Blue-Sepharose CL 6B, Sepharose 4B, and CNBr-activated Sepharose 4B were purchased from Pharmacia LKB Biotechnology, Inc. All other chemicals used were of the highest grade commercially available.

RESULTS

HPIg and HPIIm immobilized onto CNBr-activated Sepharose 4B gel were used to study the interaction with SK. IHPIg became amidolytically active as equal molar of SK was added to the solution (Fig. 1). The SK was separated from the gel of iHPIg (>97%) by washing with activating buffer followed by acid solution as monitored with 125 I-labeled SK (Table 1). When the SK moiety had been separated from the iHPIg-SK, the gel retained about 30% of amidolytic activity (Fig. 1). The amidase activities of the iHPIg-SK complex, acid-washed iHPIg-SK complex, and the iHPIIm are summarized in Table 2. The K_m of iHPIIm (746 μ M) was about twice that of the soluble HPIIm (355 μ M). The k_{cat} of iHPIIm and HPIIm were similar (23.3 and 27.9 s $^{-1}$, respectively). However, iHPIg-SK complex had lower catalytic activity. The k_{cat}/K_m of the complex was lower than that of iHPIIm (Table 2). The acid-washed iHPIg-SK had similar K_m but a smaller k_{cat} compared to that of iHPIg-SK.

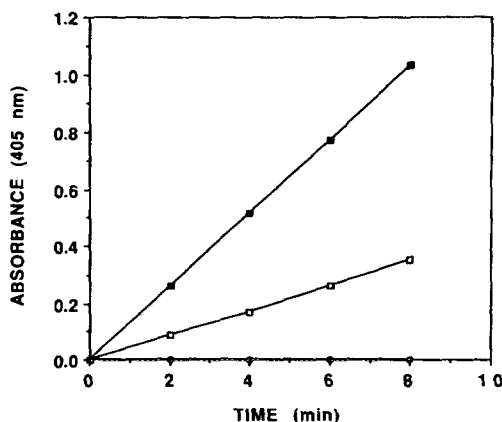


Fig. 1. Amidolytic activity of the virgin enzyme derived from the iHPIg. The amidolytic activities of 0.24 μ M iHPIg (•), iHPIg-SK before (■) and after (□) washing extensively with acid buffer are shown.

The iHPIm-SK complex was amidolytically active and could activate BPIg (Fig. 2A and 2B). After separation of SK from iHPIm-SK by acid buffer, the immobilized gel lost the ability to activate BPIg but retained its amidolytic activity (Fig. 2A and 2B). In contrast, iHPIg-SK complex was amidolytically active but could not activate BPIg (Fig. 3A and 3B). The iHPIg, after incubating with equal molar SK and soluble HPIg, exhibited BPIg activator activity in the presence of SK (Fig. 3B). The covalent binding of protein to Sepharose gel was stable since the amount of protein on the gel after treatment with cold acid buffer did not significantly decrease as measured by amino acid analysis (Table 3). The acid buffer treatment did not cause denaturation of HPIm or HPIg since more than 90% of the amidolytic activity of iHPIm was recovered after the acid wash. iHPIg was functionally active before and after acid wash as it could be activated by urokinase or equimolar HPIm-SK complex (Fig. 4). In reaction with equimolar iHPIm, SK was cleaved and major peptide fragments of 40 KDa, 36 KDa, 34 KDa and 30 KDa were found in the reaction mixture at 10 min incubation as shown by SDS gel electrophoresis. After 30 min, SK was further cleaved into peptide fragments of 30 KDa and 18 KDa. However, in reaction with iHPIg, no

Table I. Interaction of iHPIm and iHPIg with 125 I-SK and soluble 125 I-HPIg

Contents of the reaction mixture	% of total radioactivity		
	pH 6.5 eluent	pH 2.2 eluent	immobilized gel
iHPIm + SK*	91.5	6.9	1.6
iHPIg + SK*	86.9	10.5	2.6
iHPIg + SK* + HPIg	97.9	1.3	0.8
iHPIg + SK + HPIg*	96.0	1.8	2.2

iHPIm (4 μ M) or iHPIg (4 μ M) was incubated with equimolar 125 I-SK (SK*) in 0.5 ml of 0.05 M phosphate, 0.1 M NaCl, 0.02 M lysine, 25% glycerol buffer (pH 6.5) at 25°C for 30 min and then treated with buffer alone, or soluble unlabeled or labeled HPIg (4 μ M) for 30 min. The immobilized gel was washed extensively with 20 ml of pH 6.5 buffer and then 20 ml of pH 2.2 buffer at 4°C. Eluents and the remaining immobilized gel were collected to measure the radioactivities with a LKB γ -counter.

Table 2. Amidase parameters of HPIIm, iHPIIm, and iHPIg-SK complex with S-2251 at pH 7.4 and 37°C

Enzyme species	Amidase parameters		
	K_m , μM	k_{cat} , s^{-1}	k_{cat}/K_m , $\mu\text{M}^{-1}\cdot\text{s}^{-1}$
HPIIm	355 ± 3	27.9 ± 0.3	0.078
iHPIIm	746 ± 8	23.3 ± 0.8	0.031
iHPIg-SK	1151 ± 15	5.7 ± 0.6	0.005
Acid-washed iHPIg-SK	1052 ± 33	1.9 ± 0.1	0.0018

Values given are means \pm S.E. (n=4).

peptide bond cleavage of SK was observed, and only intact SK molecule was detected in the reaction mixture (Fig. 5).

DISCUSSION

SK has been used as a thrombolytic agent for decades because of its specific interaction with HPIg or HPIIm to form a complex which can activate Plg from different species. In the HPIg and SK reaction, an equimolar complex was detected (7, 8, 10, 32). The complex was rapidly converted to HPIIm-SK' (modified SK) (22-24). It was also suggested that HPIg-SK was a more effective activator than HPIIm-SK (8, 20). In the interaction of HPIg and SK, a virgin enzyme was observed (24). However, in the BPIg activation process catalyzed by HPIg-SK complex, the complex was quickly converted to HPIIm-SK' (22-24). Therefore, it was difficult to conclude which form of HPIg and SK complex was the major activator that catalyzed the conversion of BPIg to BPIIm. It was also suggested that HPIg-SK could be intramolecularly converted to HPIIm-SK' (21-23). However, no unequivocal evidence was provided to support whether the conversion was a unimolecular reaction or was catalyzed by second HPIg or HPIIm molecule.

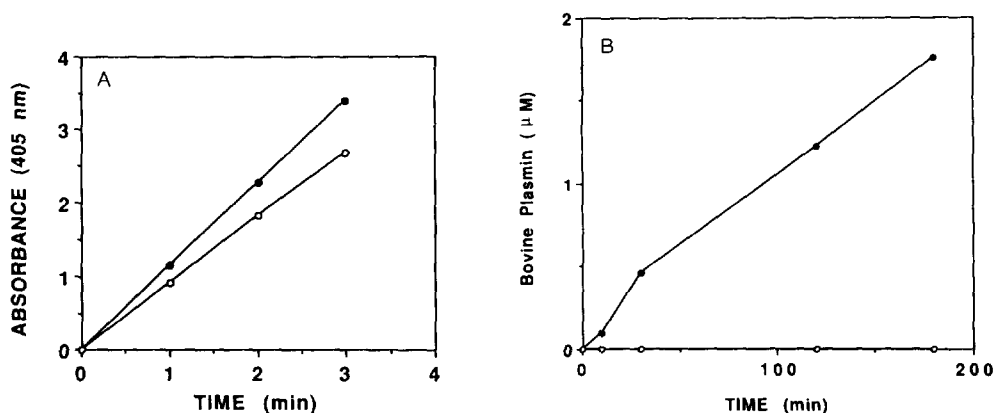


Fig. 2. The amidolytic activity (A) and BPIg activator activity (B) of iHPIIm-SK complex and the effect of washing with acid buffer. The amidolytic activity was assayed by mixing substrate S-2251 (0.5 mM) with equimolar complex of iHPIIm and SK (0.24 μM), before (•) and after (○) washing with acid buffer. The same gel samples (0.24 μM) were assayed for BPIg activator activity by mixing the gel suspension with BPIg (5 μM); the solution was assayed for BPIIm produced at determined time periods based on a BPIIm calibration curve.

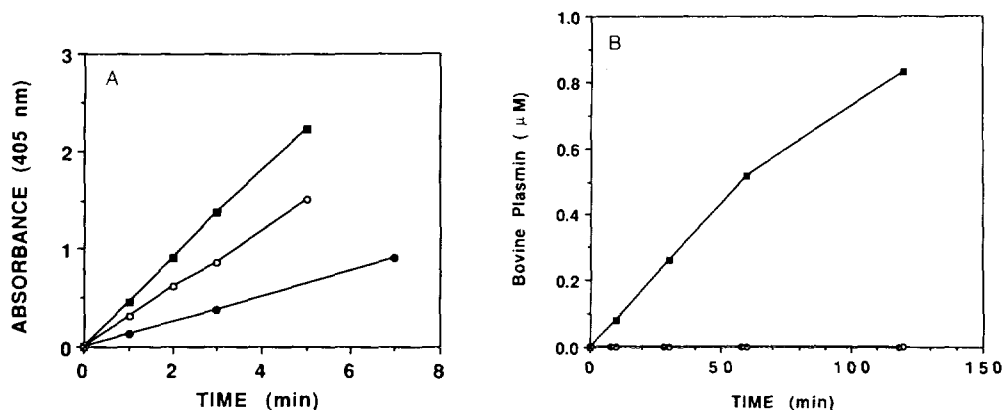


Fig. 3. The amidolytic activity (A) and BPIg activator activity (B) of the iHPIg derivatives. iHPIg (4 μ M) was incubated with SK (4 μ M) and soluble HPIg (4 μ M) for 30 min at 25°C and washed free of SK and HPIg with cold buffers of pH 6.5 and 2.2, subsequently. The immobilized gel was then resuspended in 0.5 ml of pH 6.5 activating buffer with SK (4 μ M). The amidolytic activity (A) and BPIg activator activity (B) of iHPIg-SK (●), iHPIg after interaction with SK and HPIg and after washing (○), and the mixture of SK plus the former HPIg-SK treated iHPIg (■) were measured as in previous figure.

In this study, iHPIg was used to study the conversion of HPIg-SK to virgin enzyme and to HPIg-SK. Interaction of SK with iHPIg gel suspension was limited to the reaction in which only one HPIg molecule was involved, since the interaction among HPIg on the solid support was limited. Therefore, iHPIg and iHPIg was applied to differentiate the reaction either with one or more than one HPIg involved.

The iHPIg proved to be functionally intact since it could be activated by urokinase or HPIg-SK complex, and the activated iHPIg had similar amidolytic activity as iHPIg (Fig. 2A, 3A). The k_{cat} of the activated iHPIg was about half that of iHPIg. This could be due to the steric hindrance that might have interfered with the activation of some of the HPIg molecules on the gel. The iHPIg and iHPIg had similar ability to form complex with SK; 7 to 10% of SK was retained by the gel even after washing with large volume of buffered solution. The complex was unstable and dissociated in pH 2.2 buffer.

iHPIg-SK complex was amidolytically active (Fig. 1). The HPIg on the gel remained amidolytically active as SK moiety was removed. The iHPIg-SK complex was different from the iHPIg-SK in three respects. First, the complex of iHPIg-SK was less active amidolytically than the iHPIg or iHPIg-SK complex. The k_{cat}/K_m was about 1/15 of HPIg (Table 2).

Table 3. Protein amount of iHPIg before and after washing with acid buffer

	Amount of PIg (nmole)
Control	3.04 \pm 0.04
Acid-washed	2.87 \pm 0.03

Values given are means \pm S.E. (n=4)

iHPIg was packed into a small plastic column and washed extensively with 20 ml of cold activating buffer (pH 6.5) followed by 20 ml of cold 0.05 M phosphate, 0.1 M NaCl, 0.02 M lysine, 25% glycerol buffer (pH 2.2). The immobilized gel was then washed with 30 ml of water and amount of protein was determined by amino acid analysis.

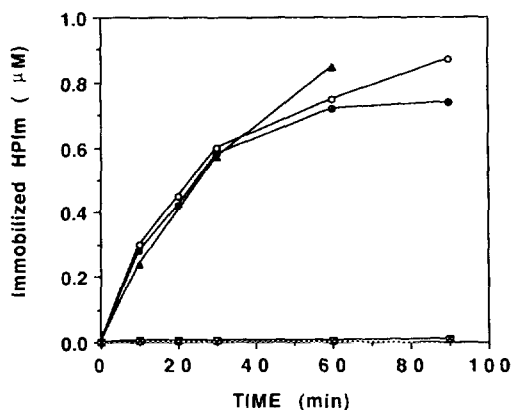


Fig. 4. Activation of iHPIg by urokinase, HPIg-SK, and SK and the effect of washing with acid buffer. Samples of iHPIg (3 μ M), before (closed symbol) and after (open symbol) washing with acid phosphate buffer was activated by different activators. Samples of gel suspension were mixed with urokinase (0.06 μ M, ●, ○), HPIg-SK (0.06 μ M, ▲), and SK (0.06 μ M, ■) in activating buffer (pH 6.5) at 30°C. HPIg-SK complex was prepared by premix equimolar concentration of HPIg and SK at 25°C for one min before adding to the reaction mixture. Samples were withdrawn at the times indicated, and amidolytic activity was assayed. The amount of iHPIg was determined on a iHPIg calibration curve.

Second, the SK moiety in interaction with iHPIg remained intact but was cleaved rapidly to fragments in interaction with iHPIg (Fig. 5). Third, the iHPIg-SK but not iHPIg-SK' catalyzed the activation of BPIg. Therefore, iHPIg was not converted to iHPIg-SK or iHPIg-SK' in interaction with SK under the experimental conditions. The amidolytically active iHPIg could be a virgin enzyme with intact Arg⁵⁶⁰-Val⁵⁶¹ peptide bond which has been characterized in soluble form of HPIg-SK complex (21). When the HPIg as well as SK were added to the iHPIg

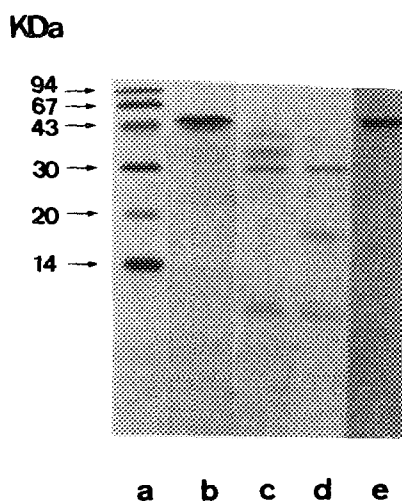
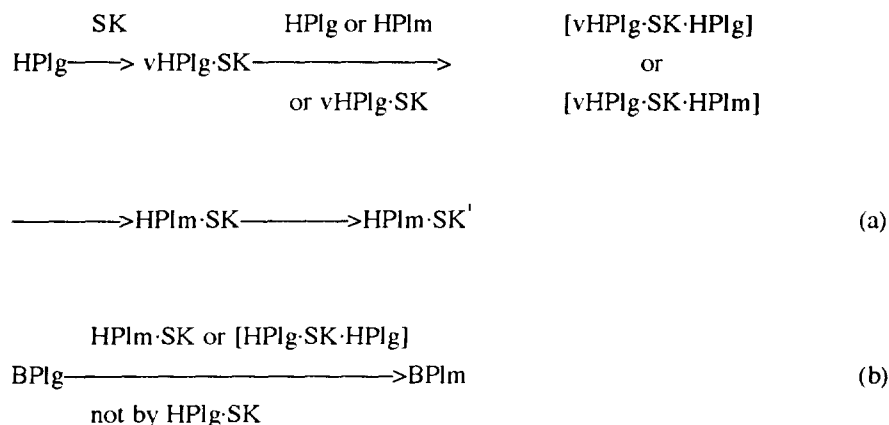


Fig. 5. SDS-gel electrophoresis of the interaction of iHPIg and HPIg with SK. Small portions of the reaction mixture supernatant of iHPIg (4 μ M) and SK (4 μ M) were taken for SDS gel electrophoresis after incubation at 25°C for various time interval as indicated: 0 (lane a), 10 (lane b), and 30 (lane c) min. Similar incubation of iHPIg with SK for 30 min (lane d) is shown.

suspension, the iHPIg was apparently converted to iHPIIm. The activated iHPIg had similar catalytic activity either as amidase or BPIg activator as iHPIIm. Therefore, SK alone cannot activate iHPIg probably because the activation of HPIg by SK required second molecule of HPIg or HPIIm for the conversion of HPIg to HPIIm. The reaction process that took place could be as follows:



In the proposed mechanism, the HPIg-SK complex was converted to the virgin enzyme complex, vHPIg-SK, through conformational rearrangement. The intramolecular activation of vHPIg-SK to HPIIm-SK was not observed in this study. The interaction with second molecule of HPIg or HPIIm was required to form a trimeric intermediate for the conversion of HPIg to HPIIm shown in the reaction scheme (a). The iHPIIm-SK can activate BPIg by direct hydrolysis of Arg⁵⁵⁷-Ile⁵⁵⁸ peptide bond. However, iHPIg-SK could not activate BPIg (b). Kinetics studies of PIg activation by native and mutant HPIg-SK complexes in solution suggested that vHPIg-SK complex in solution was a better activator for BPIg than HPIIm-SK complex (8, 20). The discrepancy could be due to the fact that the HPIg and SK complex in solution might have existed as HPIg-SK-HPIg or (HPIg-SK)₂ intermediates which could be the effective activator of both BPIg and HPIg. The trimer structure analogue was reported in the mixture of HPIg, SK and HPIIm-B chain (24). This kind of complex might not form with iHPIg, SK and BPIg, since no strong interaction of BPIg and SK was observed. Therefore, BPIg could not be activated by interaction with iHPIg-SK. The observation that HPIg-SK complex had higher PIg activator activity than HPIIm-SK complex in previous report (8, 20) was probably due to the instability of SK in interaction with HPIIm. In reaction with HPIIm, the SK moiety was probably hydrolyzed to 30 KDa and 18 KDa fragments that had less PIg activator activity compared to the intact SK (33 and Fig. 5). Further studies to characterize the structure and function of SK and PIg complexes would clarify the mechanism of PIg activation by SK.

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